Title: Influence of transboundary transport of trace elements on mountain peat geochemistry (Sudetes, Central Europe)

Author: Barbara Fiałkiewicz-Koziel, Edyta Łokas, Mariusz Gałka, Piotr Kołaczek, Francois De Vleeschouwer, Gael Le Roux, Beata Śmieja-Król

Influence of transboundary transport of trace elements on mountain peat geochemistry (Sudetes, Central Europe)

Barbara Fialkiewicz-Kozielska a,⁎, Edyta Łokas b, Mariusz Gałka c, Piotr Kołaczek a, François De Vleeschouwer d, Gael Le Roux e, Beata Smieja-Król f

a Adam Mickiewicz University, Faculty of Geographical and Geological Sciences, Krygowskiego 10, Poznań, Poland
b Institute of Nuclear Physics, Polish Academy of Sciences, Radzikowskiego 152, 31-342, Kraków, Poland
c University of Lodz, Faculty of Biology and Environmental Protection, Department of Geobotany and Plant Ecology, Banacha 12/16, Lodz, Poland
d Institute Franco-Argentino para el Estudio del Clima y sus Impactos (UMI IFAECI/CNRS-CONICET-UBA), Universidad de Buenos Aires, Intendente Guiraldes 2160, Ciudad Universitaria, Pabellon II - 2do. Piso, (C1428EGA) Ciudad Autonoma de Buenos Aires, Argentina
e Ecolab, Université de Toulouse, CNRS, INPT, UPS, Toulouse, France
f University of Silesia, Institute of Earth Sciences, Faculty of Natural Sciences, Będzinśka 60, 41-200, Sosnowiec, Poland

⁎ Corresponding author.
E-mail address: basiafk@amu.edu.pl (B. Fialkiewicz-Kozielska).

1. Introduction

Industrial activities caused dramatic changes in ecosystems, releasing uncontrolled amounts of trace elements influencing both biotic and abiotic components (Rose, 2015; Waters et al., 2016, 2018).

Atmospheric pollutants are transported by wind and dispersed at different geographical scales. The range of dispersion is dependent on the size of the particle and its chemical behaviour in the air (Samson, 1988). Larger particles are generally deposited close to the emitter (Samson, 1988) while regionally-recognized lighter particles can be transported from hundreds to thousands of kilometres, passing through cities, states, and countries, and strengthen the local pollution (Munn, 1972; Erel et al., 2002, 2006; Bergin et al., 2005). It results in a more global, uniform pollutant signal over continents for certain pollutants (i.e. contaminants carried by submicronic aerosols: artificial radionuclides, lead derived from the combustion of leaded gasoline).

Mountain ecosystems such as lakes and peatlands are sensitive archives of long-range transport (Arellano et al., 2011; Catalán et al., 2013; Le Roux et al., 2016; Okamoto and Tanimoto, 2016). Notwithstanding, the received input of dust depends on height above sea level (a.s.l.). The altitudes between 0 and ~2000 m a.s.l. are mainly subjected to vertical transport of Potentially Harmful Trace Elements (PHTE) (Le Roux et al., 2016). The free tropospheric zone, located above 2000 m a.s.l. is characterised by more lateral
transport of PHTE (Catalan et al., 2013; Le Roux et al., 2016) and ecosystems located there receive dust mostly from long-range transport. The mutual effect of different climate factors like wind trajectories and precipitation play the most important role in mountain pollution (Catalan et al., 2013; Le Roux et al., 2016).

Peat bogs, exclusively fed by dry and wet atmospheric deposition, are significant biological filters and traps for PHTE and other elements. They give essential information about past and present changes in the environment, especially when applying a multiproxy approach (Lamentowicz et al., 2013; Galka et al., 2019). Mountain peat bogs, located on plateaus, where the slope effect is minimal, are great archives of geochemical records of various origin (Le Roux et al., 2016).

The calculated accumulation of peat and, therefore, accumulations of elements enable to assess the past trends in pollution as well as to assess the sources of observed enrichments (e.g. Shotyk et al., 1998, 2002). REE elements are believed to be good indicators of lithogenic activity and natural sources of dust (e.g. Le Roux et al., 2012; Pratte et al., 2017; Vanneste et al., 2016), but they can also be indicators of anthropogenic activity (e.g. Fiałkiewicz-Koziel et al., 2016).

The Sudetes (located at the border of Poland, the Czech Republic, and Germany) are an important region with long-time traditions of mining and ore processing (i.e. Fe, Cu, Pb, U, and Au) that may have had a great effect on PHTE deposition onto mountains. This mountain range is additionally surrounded by economically significant industrial centres and lignite-fired power plants, causing dramatic deterioration of environment in the 1970s and 1980s in Poland, the Czech Republic, and Germany. Due to dramatic environmental impact this area (e.g. Mazurski, 1986), named “Black Triangle” (BT), was qualified by the United Nations Environment Programme (UNEP) as an “ecological disaster zone” (Grübler, 2002; Kolár et al., 2015). Sparse investigations about mineralogy, pollen analysis, sulfur and carbon isotopes exist (i.e., Popowski, 2005; Skrzypek et al., 2009; Kajukuio et al., 2016), but information about past PHTE contamination in peat bogs from the Sudetes is scarce (i.e. Strzyszcz and Magiera, 2001; Fiałkiewicz-Koziel et al., 2015a). The most detailed anthropogenic signal in central Europe was estimated in the Czech Republic (i.e., Novak et al., 2003, 2008; Ettler et al., 2004; Ettler et al., 2006; Mihalic, 2006; Bohdoalova et al., 2014). Only a few multiproxy studies including advanced geochemical analyses were carried out on peatlands from Poland (e.g. De Vleeschouwer et al., 2009; Fiałkiewicz-Koziel et al., 2018).

Here we present a detailed multiproxy geochemical record from two cores of the Na Równy pod Śnieżką (NRS) peatland, located at the feet of the Śnieżka summit (1400 m a.s.l.), the highest mountain of the Sudetes. We hypothesised that long-range transport mainly contributes to the pollution of NRS. Our aims are: (1) to assess the pattern of distribution and origin of trace elements and REE (2) to distinguish the most significant signals of pollution in areas impacted by transboundary anthropogenic pollution and (3) to explain the fate of strategic elements like uranium after anthropogenically-driven release to the atmosphere.

2. Material and methods

2.1. Study site

The Sudetes are surrounded by industrial centres located in Poland (Lower Silesia) as well as in the Czech Republic (Northern Bohemia) and Germany (Saxony and Brandenburg). These three areas constitute the so-called Black Triangle (Fig. 1). In the 1970s and 1980s, the “Black Triangle” was one of the most polluted areas in Europe due to the uncontrolled density of factories and power plants, which, working without filters, released enormous amounts of alkali dust and toxic gases (mainly SO2) in the atmosphere (Jedrysek et al., 2002; Sznukiewicz et al., 2008).

The Na Równy pod Śnieżką (NRS) peatland (50° 44′18.31″N, 15° 41′59.72″E) is located on a plateau (1350–1450 m a.s.l.) in the Karkonosze range, close to the highest summit of Mt Śnieżka (1602 m a.s.l.). The Karkonosze area is –60 km long and up to 20 km wide granitic massif in the West Sudetes at the northern periphery of the Bohemian Massif, straddling the Czech/Polish border and forming the eastern extremity of the European Variscan belt (Mazur et al., 2006). The Karkonosze unit was lifted during Alpine orogeny, forming currently a steep rocky mountain range with numerous glacial landforms. Consequently, it is a significant orographic barrier for different types of air masses. During the 64% of the year, the area is directly influenced by the zonal circulation of oceanic air masses from the North Atlantic, flowing over the lowlands of Western Europe. 30% of the year is characterised by the influence of polar continental air masses, flowing from the east, 4% by arctic air from the north and 2% by tropical air masses, coming from the south (Sobik et al., 2014). The mean annual air temperature at Mt. Śnieżka is +0.5 °C and the annual precipitation ~1500 mm (Sobik et al., 2014; Migala et al., 2016). The dominant winds are the westerlies, amongst which SW and S winds dominated while NW and W are less frequent (Schwartz et al., 1994; Sobik et al., 2014, Fig. 1).

2.2. Coring and subsampling

Two cores of 63 and 67 cm long (Sn1 and Sn2, respectively), were retrieved in the spring of 2012 using a stainless steel 10 × 10 × 100 cm Wardenaar corer (Wardenaar, 1987). The distance between them was 20 m. Monoliths were wrapped in plastic bag, transported to the laboratory in Poznań (UAM), and stored in the laboratory cooler. Fresh cores were cut into 1-cm thick slices (except the top 6 cm, which were cut on 2 cm slices) using a carbon steel knife. Fresh cutting was necessary for detailed mineralogical analysis as freezing the peat core (as it is frequently done before slicing) would have caused a transformation of minerals by changing the temperature and water conditions of the catotelm. Protocols from Givel et al. (2004) and De Vleeschouwer et al. (2010) were used to minimise contamination. Several geological samples such as the local rocks (granite, hornfels from Śnieżka), lignite coal (mine Turosów), fly ash from Power Plant Turów, aircraft fuel (Laboratory Warter fuels, Plock, Poland) were also collected.

2.3. Ash content, bulk density, plant macrofossils

Dried bulk density was determined using fresh material collected with a 5-cm3 beaker. Ash content (AC), used to quantify the relative proportion of mineral fraction in the peat, was determined by burning the dry samples at 550 °C overnight. High-resolution (1-cm peat slices) plant macrofossil analysis was used to reconstruct local ecological conditions and peat-forming plants in contiguous samples of approximately 20 cm2 in the two profiles (comp. Suppl.1). The samples were washed and sieved under a warm water current over 0.2-mm mesh sieves. Vascular plants and brown mosses composition were determined on carpospores using spe- cific keys (Hölzer, 2010; Laine et al., 2011).
2.4. Dating

The Sn1 profile was analyzed by gamma ($^{137}\text{Cs}$) and alpha ($^{210}\text{Po}$, $^{238}\text{Pu}$, $^{239-240}\text{Pu}$, $^{233}\text{U}$) spectrometry (comp. Table 1, Fig. 2). Separated slices were dried at 105°C, homogenised, and taken for gamma spectrometric measurement. For $^{137}\text{Cs}$ analyses, samples were packed into 100 ml polypropylene cylindrical containers and were measured using high-resolution gamma spectrometry with a planar HPGe (high-purity germanium) detector with a composite foil window made of carbon fiber which support a capon foil covered with ultra-thin aluminum foil (homemade by Institute of Nuclear Physics PAS Krakow and electronics by Silena S.p.A.). A small amount of sample material was deposited on the bottom of the container. Activities of $^{137}\text{Cs}$ were determined via the $^{137m}\text{Ba}$ emission peak at 662 keV. Spectra were collected for 12–72 h, depending on the activity of the samples. The activities of $^{210}\text{Po}$, $^{238}\text{Pu}$, $^{239-240}\text{Pu}$, and $^{233}\text{U}$ were determined for about 1 g of dried samples. Samples were digested with radioactive tracers ($^{208}\text{Po}$, $^{242}\text{Pu}$, $^{232}\text{U}$) and a concentrated mixture of HNO$_3$, HCl, H$_3$BO$_3$, and H$_2$O$_2$ and slowly dried. Details of this procedure were described in previous articles (Mietelski et al., 2008; Łokas et al., 2013; Mróz et al., 2017). The activity of total $^{210}\text{Pb}$ was determined indirectly by measuring its decay product, $^{210}\text{Po}$, using alpha spectrometry. $^{210}\text{Po}$ was chemically extracted from the material. Po isotopes were deposited on an Ag disc. Solution after polonium separation was used for Pu separation. Plutonium alpha sources were prepared by the NdF$_3$ microcoprecipitation method (Sill, 1987; Rao and Cooper, 1995). The effluent solution (8 M HNO$_3$) after Pu separation was used for U determination. Uranium was co-precipitated directly from this elution using NdF$_3$ and Mohr’s Salt (ammonium iron sulfate) to obtain a thin spectrometric source (Łokas et al., 2010; Mietelski et al., 2016). Activity concentrations for polonium, plutonium, and uranium isotopes were determined using alpha spectrometers (Silena AlphaQuattro, Ortec Alpha Duo or Canberra 7401; all equipped with Canberra or Ortec ion-implanted silicon detectors). For this radiochemical procedure blank samples were made of reagents and were analyzed after each thirty sample. Alongside blanks and samples, the reference materials (IAEA Moss-Soil 447 and IAEA Sediment 385) were analyzed to ensure the quality of measurements. The obtained results: 447 ($^{234}\text{U}$ – 20.8 ± 1.8; $^{238}\text{U}$ – 21.4 ± 2.2; $^{239-240}\text{Pu}$ – 0.14 ± 0.01; $^{233}\text{U}$ – 4.96 ± 0.32, $^{210}\text{Po}$ – 415 ± 10; $^{137}\text{Cs}$ – 421 ± 22); 385 ($^{234}\text{U}$ – 27.9 ± 1.9; $^{238}\text{U}$ – 29.4 ± 2.2; $^{239-240}\text{Pu}$ – 0.40 ± 0.06; $^{233}\text{U}$ – 2.70 ± 0.23) fall within 93–99% of certified values.

Seven samples from Sn1 and five samples from Sn2 were subjected to $^{14}\text{C}$ measurements in Radiocarbon Laboratory in Poznań. Hand-picked plant macrofossils, stored in MilliQ water, were selected for dating (Table 2). An absolute chronology is based on (i) the age-depth models calculated on $^{14}\text{C}$ dates in the OxCal v. 4.3 software (Bronk Ramsey, 1995) applying the IntCal13 (Reimer et al., 2013) and BOMB13NH1 (Hua et al., 2013) atmospheric curves as the calibration set and (ii) $^{210}\text{Pb}$ dates obtained for the section of 0–52 cm in the Sn1 profile (Fig. 3). Additionally, patterns of SAP and mullite were used as chronomarkers and included in age-depth model for Sn2 (comp. Fiałkiewicz-Kozieł et al., 2016). For the calculation of the models based on $^{14}\text{C}$ dates, the $P_{\text{Sequence}}$ command (Bronk Ramsey, 2008) with k parameters equal to 1 cm$^{-1}$ and log$_{10}(k/k_0)$ = 1 was applied. In the case of the Sn1 profile, the section between 0 and 52 cm, being dated using $^{210}\text{Pb}$ method (CRS model) (Table 1 and Fig. 2), was validated by $^{14}\text{C}$ dates. The CRS model assumes a constant rate of supply of unsupported $^{210}\text{Pb}$ to the peat surface despite variable sedimentation rates (Appleby and...
Unsupported activity concentrations of $^{210}$Pb, which were determined for each layer by subtracting the supported activities from the total $^{210}$Pb activities. The supported level of $^{210}$Pb was calculated by using the mean activity of the bottom layers of the peat profile (14 ± 2 Bq kg$^{-1}$).

The lower section of the profile (52–64 cm) was based on the

Oldfield, 1978). Unsupported activity concentrations of $^{210}$Pb, which were determined for each layer by subtracting the supported activities from the total $^{210}$Pb activities. The supported level of $^{210}$Pb was calculated by using the mean activity of the bottom layers of the peat profile (14 ± 2 Bq kg$^{-1}$).

The lower section of the profile (52–64 cm) was based on the
Bayesian age-depth model constructed using the mixed $^{210}$Pb (date from a depth of 52.5 cm introduced using C_Date command) and $^{14}$C dates. In the case of the Sn2 model, the calendar date AD 1938 ± 2 was introduced (using C_Date command) at a depth of

### Table 2

Radiocarbon dates of Sn1 and Sn2 profiles. Abbreviations: pMC – percent of modern $^{14}$C.

<table>
<thead>
<tr>
<th>Depth [cm]</th>
<th>Material dated</th>
<th>Nr. Lab.</th>
<th>$^{14}$C date (AMS)</th>
<th>Calibrated date (range 95.4%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sn-1</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>35–36</td>
<td>Stems with leaves of Polytrichum strictum</td>
<td>Poz-53963</td>
<td>100.66 ± 0.31 pMC</td>
<td>1952–1957 (3.1%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1990–1996 (91.3%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>2008–2008 (1%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1682–1735 (27.5%)</td>
</tr>
<tr>
<td>40–41</td>
<td>Stems with leaves of P. strictum</td>
<td>Poz-53964</td>
<td>105 ± 25</td>
<td>1695–1728 (22%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1806–1930 (67.9%)</td>
</tr>
<tr>
<td>45–46</td>
<td>Stems with leaves of P. strictum</td>
<td>Poz-53965</td>
<td>60 ± 25</td>
<td>1812–1854 (20.1%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1857–1919 (53.3%)</td>
</tr>
<tr>
<td>50–51</td>
<td>Stems with leaves of Calliergon stramineum</td>
<td>Poz-53966</td>
<td>100 ± 30</td>
<td>1682–1736 (27.1%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1805–1935 (68.3%)</td>
</tr>
<tr>
<td>55–56</td>
<td>Stems with leaves of C. stramineum</td>
<td>Poz-53967</td>
<td>170 ± 30</td>
<td>1659–1699 (17.3%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1721–1818 (50.5%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1833–1880 (83%)</td>
</tr>
<tr>
<td>60–61</td>
<td>Stems with leaves of C. stramineum</td>
<td>Poz-53968</td>
<td>200 ± 30</td>
<td>1646–1690 (24.9%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1729–1810 (51.2%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1916–modern (19.6%)</td>
</tr>
<tr>
<td>63–64</td>
<td>Stems with leaves of C. stramineum</td>
<td>Poz-53969</td>
<td>120 ± 25</td>
<td>1680–1764 (31.4%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1801–1939 (64%)</td>
</tr>
<tr>
<td>Sn-2</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>11–12</td>
<td>Stems with leaves of P. strictum</td>
<td>Poz-66397</td>
<td>106.83 ± 0.35 pMC</td>
<td>1956–1957 (3.1%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>2002–2007 (91.3%)</td>
</tr>
<tr>
<td>27–28</td>
<td>Stems with leaves of Sphagnum russowii</td>
<td>Poz-66399</td>
<td>200 ± 30</td>
<td>1646–1690 (24.9%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1729–1810 (51.2%)</td>
</tr>
<tr>
<td>43–44</td>
<td>Stems with leaves of S. russowii/lindbergi</td>
<td>Poz-55190</td>
<td>120 ± 35</td>
<td>1677–1765 (33.4%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1773–1777 (0.8%)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1800–1940 (61.2%)</td>
</tr>
<tr>
<td>52–53</td>
<td>Stems with leaves of Drepanoclados sp.</td>
<td>Poz-55191</td>
<td>960 ± 50</td>
<td>990–1186 (95.4%)</td>
</tr>
<tr>
<td>63–64</td>
<td>Stems with leaves of S. russowii</td>
<td>Poz-56357</td>
<td>1710 ± 35</td>
<td>246–401 (95.4%)</td>
</tr>
</tbody>
</table>

**Fig. 3.** Age-depth models of both profiles. SAP – spheroidal aluminosilicates.
2.5. Pb, Zn, Cu, Ni, Cr, Sr, Ti, Al, and REE concentrations

Peat samples were dried at 105 °C prior to ashing at 550 °C and acid digestion in Teflon Savillex® beakers. Acid digestion consisted of two steps. First, samples were treated with 4.5 ml of suprapure 65% HNO₃ and 0.5 ml of 45% HF for 48 h at 130 °C. They were then evaporated and treated with 1 ml of concentrated HCl for 24 h in 90 °C. After final evaporation, residues were diluted in 10% HNO₃ and measured using ICP-MS at the UAM for trace metals and at Observatoire Midi-Pyrénées (Toulouse, France) for Rare Earth Elements. All measurements were performed in duplicate. NIMT (peat, Yafa et al., 2004) and CTA-OTL-1 (tobacco leaves) certified reference materials were used to monitor the accuracy of measurements. Certified value of NIMT equals: Al - 3692 ± 347 mg kg⁻¹; Cr - 6.36 ± 0.44 mg kg⁻¹; Cu - 5.28 ± 1.04 mg kg⁻¹; Fe - 921 ± 84 mg kg⁻¹; Pb - 174 ± 8 mg kg⁻¹; Ti - 357 ± 18 mg kg⁻¹; Zn - 28.6 ± 1.9 mg kg⁻¹; La - 1.24 ± 0.03 mg kg⁻¹; Sm - 0.194 ± 0.005 mg kg⁻¹. Measured CRM values fall within 97% of the certified values for Al, Ti, Cr, Cu, Fe, Pb, Zn, and 85–86% for La and Sm. The analytical deviation is less than 5%.

The accumulation rate (AR) was calculated using concentration (µg g⁻¹) of element, bulk density, and peat accumulation rate using the following equation:

\[
\text{AR (mg m}^{-2} \text{yr}^{-1}) = \text{concentration} \times \text{BD (g cm}^{-2}) \times \text{PAR (cm y}^{-1})
\]

The dust flux was calculated using the sum of REE concentrations (µg g⁻¹) in the bulk peat using the following equation (Shotyk et al., 2002):

\[
\text{Dust flux (gm}^{-2} \text{yr}^{-1}) = \left( \frac{\sum \text{[REE]}_{\text{sample}}}{\sum \text{[REE]}_{\text{UCC}}} \right) \times \text{PAR (cm y}^{-1}) \times \text{BD (g cm}^{-3}) \times 10,000
\]

were \(\Sigma [\text{REE}]_{\text{sample}}\) is the sum of REE concentrations (µg g⁻¹) in a sample, \(\Sigma [\text{REE}]_{\text{UCC}}\) is the sum of REE concentrations in the upper continental crust (144.3 µg g⁻¹; Wedepohl, 1995), PAR is the peak accumulation rate (cm yr⁻¹) calculated according to formula (h₂-h₁)/(y₁-y₂), where h-depth, y-age, and BD is the bulk density (g cm⁻³).

2.6. Pb isotopes

Ten to 500 mg of dry peat and rocks powder was taken in order to obtain 2000 ng of Pb in the final solution. The sample powders were ashed (550 °C, 4 h) prior to digestion in 6 ml HF suprapur (Merck) + 1 ml HNO₃ 65% cc. sub. in Teflon beakers (120 °C, 48 h) in class 1000 clean room (UAM, Poznan). Dried residues were dissolved in HBr 6% cc. sub. prior to chromatographic separation. Lead was separated using anionic exchange micro-columns (Weis et al., 2005) and sub-boiled distilled acids. The measurements were conducted on TIMS Finnigan MAT-261 special (UAM). The instrumental drift was controlled by standard bracketing using NBS981 standard data (Galer and Abouchami, 1998). Repeated standard measurements for NBS981 equal: \(208/204\text{Pb} = 36.756 ± 0.013, 207/204\text{Pb} = 15.484 ± 0.018\) and \(206/204\text{Pb} = 16.937 ± 0.010\) and are comparable with recommended values:

\(208/204\text{Pb} = 16.9406 ± 0.0003; 207/204\text{Pb} = 15.4957 ± 0.0002, 206/204\text{Pb} = 36.7184 ± 0.0007\) (Taylor et al., 2015).

2.7. Mineralogy

The shape, size, morphology, and chemical composition of dust particles were determined in peat samples using the backscattered electron detector of a scanning electron microscope (SEM) equipped with an energy dispersive system (Philips XL30 ESEM/EDS). The accelerating voltage was 15 kV and 10 mm the working distance. Air-dried peat samples were gently homogenised using a corundum mortar and pestle. A thin layer of each homogenised sample was fixed to a double-sided 9 mm carbon tab, placed on an aluminium stub, and carbon-coated prior to analysis.

Mullite content was determined in Sn1 profile using X-ray diffraction (XRD). For the XRD analysis, peat samples were ashed at 550 °C and treated with 1 M HCl for 15 min to remove the acid-soluble ash fraction (Sapkota, 2006). The residue was dried, ground in an agate mortar, and analyzed using a Panalytical X’Pert PRO - PW 3040/60 X-ray diffractometer. The instrument was equipped with a Ni-filtered Cu Kα source radiation (\(\lambda = 1.540598 \text{Å}\)) and an X’Celerator strip detector. Samples were scanned within a 2θ interval of 2.5–65°, with a step size of 0.01° 2θ and counting time of 300 s. Identification and quantification of mineral phases was done by means of the X’Pert HighScore Plus Software using the newest ICSD database. The detection limit of the XRD method was 0.5–2%, depending on the sample mineral composition. Analytical precision and accuracy were ±3%.

3. Results

3.1. Chronological control (\(^{210}\text{Pb}, ^{14}\text{C}\))

The \(^{210}\text{Pb}\) activities (Table 1) gradually declined with increasing depth and became constant at depths ranging from 52.5 to 54.5 cm, which was consistent with the equilibrium depth of total \(^{210}\text{Pb}\) and supported \(^{210}\text{Pb}\) (14 ± 2 Bq kg⁻¹). Fig. 2 shows the CRS age-depth relationships of Sn1 peat profile. The peat section 0 – 49 cm accumulated in 149 years, corresponding to an average accumulation rate of 0.53 ± 0.01 cm yr⁻¹.

The data obtained from \(^{14}\text{C}\) activity (Table 2) suggests a hiatus in the Sn2 profile. Therefore, for further interpretation, we opted only to use the top 44 cm, which spans a similar time interval comparing to Sn1.

The \(A_{\text{model}}\) for the Sn1 model revealed 17%, whereas for the Sn2 it was 64% (Fig. 3). The minimum advised as the critical value for the age-depth model robustness is 60% (Bromk Ramsey, 2008). However, in the case of Sn1, plotting numerous \(^{210}\text{Pb}\) dates (as calendar dates), even if excluded from the modelling, caused a decrease in \(A_{\text{model}}\). If the Sn1 model is plotted without drawing the calendar dates), even if excluded from the modelling, caused a decrease in \(A_{\text{model}}\). If the Sn1 model is plotted without drawing the

--

\(206/204\text{Pb} = 36.756 ± 0.013, 207/204\text{Pb} = 15.484 ± 0.018, 206/204\text{Pb} = 16.937 ± 0.010\)

--

\(208/204\text{Pb} = 16.9406 ± 0.0003, 207/204\text{Pb} = 15.4957 ± 0.0002, 206/204\text{Pb} = 36.7184 ± 0.0007\) (Taylor et al., 2015).
3.2. $^{137}$Cs, $^{239,240}$Pu

Activity concentrations and inventories of anthropogenic radionuclides ($^{137}$Cs, $^{239}$Pu, $^{239,240}$Pu) are presented in Table 1 and in Fig. 2. The radionuclide inventory is understood here as the activity concentration of a given radionuclide contained in the profile (Bq kg$^{-1}$) per unit surface area (kg m$^{-2}$). The activity concentrations of $^{137}$Cs range from $81 \pm 45$ to $2255 \pm 104$ Bq kg$^{-1}$. Most of the total Cs activity concentration is retained in the upper four layers (0–9 cm). Two distinct peaks of $^{137}$Cs activity can be found at 23.5 cm and 33.5 cm depth. Activity concentrations for $^{238}$Pu in this profile were 0.03 Bq kg$^{-1}$ in Sn1 and 8.2% in Sn2 (see Suppl. 2). Two maxima with higher ash content at 22.5 cm and 35.5 cm; AD 1780, 1789 (comp. Suppl. 2) can be attributed to Chernobyl (1986) and Nuclear weapon test (1963) signal (Table 1, Fig. 2).

Activity concentrations of $^{239,240}$Pu range between $0.06 \pm 0.04$ Bq kg$^{-1}$ to $22.86 \pm 1.99$ Bq kg$^{-1}$ (dry weight) with a maximum value at 33.5 cm depth. Activity concentrations for $^{239}$Pu in this profile are much lower than for $^{239,240}$Pu. The minimum value is $<0.03$ Bq kg$^{-1}$, the maximum equals $0.69 \pm 0.14$ Bq kg$^{-1}$ at the same depth as $^{239,240}$Pu (Table 1).

3.3. Discrimination of temporal variations in the deposition of Pb, Zn, Cu, Ni, Cr, Ti, Al, Sc, and REE (concentrations and accumulations)

The ash content varied from 0.5 to 6.1% in Sn1 and between 0.7 and 8.2% in Sn2 (see Suppl. 2). Two maxima with higher ash content are observed (5.5–5.1%; 55.5–53.5 cm; AD 1806–1828) and (6.1%; 30.5 cm; AD 1970) in Sn1. The general shape of ash is similar to Ti and Al concentration pattern. Bulk density (BD), used to calculate accumulation rates, varies from 0.02 to 0.11 g cm$^{-3}$ for Sn1 and from 0.03–0.10 g cm$^{-3}$ for Sn2. The last two layers of Sn2 (45.5–44.5 cm; AD 1851–1760) were characterised by a higher density — 0.14–0.17 g cm$^{-3}$ (Suppl.2).

Several shifts in concentration and accumulation of investigated elements can be distinguished.

In Sn1 the layer between 50.5 and 62.5 cm (AD 1739–1872) was characterised by the decreased Pb concentration from 218 mg kg$^{-1}$ at the bottom of the profile to twice lower value of 111 mg kg$^{-1}$ at the depth 58.5 cm (AD 1780), which slightly fluctuated till 50.5 cm (AD 1872). The calculated Pb accumulations are the highest at the depth (62.5 cm) — 20 mg m$^{-2}$ y$^{-1}$ (Fig. 4). The increase in Al and Ti concentration from 57.5 cm to 50.5 cm; (AD 1789–1872) with maximum at 55.5–53.5 cm (AD 1806–1828) is less visible in accumulation rates (comp. Fig. 4). Other elements displayed only minor fluctuations during that time.

In Sn2 the described period was represented by three samples within a depth interval of 45.5 cm–43.5 cm (AD 1760–1883). This layer was characterised by the highest density and low accumulation rates.

The depth from 50.5 cm to 35.5 cm (AD 1870–1954) in Sn1 was characterised by only small variations in both concentration and accumulation of most elements (Suppl.2, Fig. 4). In Sn2, the depth from 43.5 cm to 20.5 cm (AD 1883–1952) reflected a more complex, but comparable trend to Sn1 (Suppl.2, Fig. 4). In both profiles, an anomaly in Cr and Ni is observed within the period. In Sn1, a distinct peak in Cr (33 mg kg$^{-1}$) and Ni (15 mg kg$^{-1}$) concentrations is seen at 46.5 cm (AD 1904; Fig. 5). The similar, but broaden peak was also observed in Sn2 at a depth of 33.5 cm (AD 1912) for Al, Ti, Cr, Ni, and Cu and at 31.5 cm (AD 1922) for Cr and Ni (comp. Suppl.2, Figs. 4 and 5). Pb also revealed a small increase in concentration as well as accumulation in both profiles.

The peat layer 22.5 cm–35.5 cm (AD 1954–1988) in Sn1 and 10.5 cm–19.5 cm (AD 1955–1978) in Sn2 was characterised by the most pronounced changes in concentrations and accumulations of all elements.

Sn1 presents a substantial increase of elements at the depth 28.5–30.5 cm (1970–1975 AD): Pb (365 mg kg$^{-1}$), Al (7740 mg kg$^{-1}$), Ti (595 mg kg$^{-1}$), Sc (2.86 mg kg$^{-1}$), Zn (289 mg kg$^{-1}$), Cu (26 mg kg$^{-1}$), Fe (6231 mg kg$^{-1}$). Cr (18 mg kg$^{-1}$), Ni (13 mg kg$^{-1}$) (comp. Suppl. 2). The corresponding increase in accumulation rates was also observed: Pb AR (77 mg m$^{-2}$ y$^{-1}$), TiAR (164 mg m$^{-2}$ y$^{-1}$), ScAR (0.63 mg m$^{-2}$ y$^{-1}$), ZnAR (56 mg m$^{-2}$ y$^{-1}$), CuAR (4.8 mg m$^{-2}$ y$^{-1}$), FeAR (1858 mg m$^{-2}$ y$^{-1}$), CrAR (5.15 mg m$^{-2}$ y$^{-1}$), NiAR (2.3 mg m$^{-2}$ y$^{-1}$) (comp. Figs. 4 and 5).

In Sn2 maximum values had a broaden extend (mostly between 11.5 and 17.5 cm; AD 1960–1974) and Pb reached the maximum...
value (198 mg kg$^{-1}$) at 14.5 cm (AD 1967) together with Ni (7.9 mg kg$^{-1}$), Aluminium (5987 mg kg$^{-1}$), Ti (473 mg kg$^{-1}$), Cr (13.5 mg kg$^{-1}$), and Cu (16 mg kg$^{-1}$) had maximum values of concentration at 17.5 cm depth (AD 1960). Most elements showed then a decreasing trend from 11.5 cm upward except Fe (11,550 mg kg$^{-1}$) and Zn (109 mg kg$^{-1}$), which had a maximum at this depth level. The second maximum, visible in Sn1 around AD 1987, was undistinguishable in Sn2, probably due to lesser resolution of the Sn2 age-depth model. The corresponding accumulation rates presented more comparable pattern within the investigated depths, respectively. In Sn2 the maximum U concentration was used to assess the dust flux reaching maximum values in 1970 (Fig. 4).

3.4. Uranium concentration and isotopes

Slightly elevated values of U concentrations (up to 1 mg kg$^{-1}$) were recorded in the lower part (31.5–62.5 cm) of Sn1. Three distinct peaks appeared at 30.5 cm (AD 1970; 2.26 mg kg$^{-1}$), 28.5 cm (AD 1975; 1.54 mg kg$^{-1}$) and 25.5 cm depth (AD 1982; 1.47 mg kg$^{-1}$) (Fig. 7). Suppl. 2 presents values of uranium activity for Sn1 which ranged between 1.3 ± 0.2 Bq kg$^{-1}$ to 37.0 ± 3.8 Bq kg$^{-1}$ for $^{234}$U, 1.0 ± 0.1 Bq kg$^{-1}$ to 21.0 ± 2.8 Bq kg$^{-1}$ for $^{238}$U. The maximum value for $^{234}$U and $^{238}$U was observed at 24.5 cm depth. Others slightly lower peaks were observed at 29.5 and 48.5 cm depths, respectively. In Sn2 the maximum U concentration (106–1.24 mg kg$^{-1}$) is presented as a broad peak within 11.5–15.5 cm (AD 1965–1974) (Fig. 7).

3.5. Tool for distinguishing sources of deposited elements (Pb isotopes)

Lead isotopic signatures revealed similar age and depth patterns in both peat profiles during the last 100 years. $^{208}$Pb/$^{204}$Pb varied from 38.55978 to 38.02255 in Sn1 and from 38.5907 to 37.9157 in Sn2; $^{207}$Pb/$^{204}$Pb – from 15.70587 to 15.75680 (Sn1); 15.6728–15.5586 (Sn2); $^{206}$Pb/$^{204}$Pb – 18.39197–18.04409 (Sn1) and 18.4274–18.1165 (Sn2) (Table 3; Fig. 4). Generally, the ratio $^{206}$Pb/$^{204}$Pb decreased to less radiogenic values from the bottom to the top of the profiles and revealed similarity to the anthropogenic sources (Figs. 4 and 6).

3.6. Mineralogy

The distinguishing of mineral particles in the peat profiles gave an opportunity to further confirm the sources. Spheroidal aluminosilicate fly ash particles (SAP) were the dominant technogenic dust particles detected in the samples using SEM. In the Sn1 profile, SAP were found within a depth range of 0–40 cm and in Sn2 at depths between 0 and 27 cm. This result was used to calculate chronology (comp. 2.4 and Fig. 3). The majority of the SAP particles in the Sn1 profile were within a size range of <1–9.5 μm,
independent of the depth, which indicates the long-range transport. The mean value was 2.4 and the median - 2.0 um (n = 180). A small number of much larger (up to 50 um), less regular, and highly porous aluminosilicates were found within a depth range of 25–35 cm in Sn1.

Mullite is a constant component of the peat ash within a depth range 18–40 cm (AD 1938–1995) and its content is more elevated (9.9–13.3%) between 23 cm and 33 cm depth (AD 1962–1988). It accounts for 1.6–13.3% of the crystalline constituents of the ash. The other crystalline phases detected in the samples in decreasing amounts are as follows: quartz (53.7–91.8%), feldspars (6.8–31.1%), layered aluminosilicates (mainly muscovite and illite <0.5–9.9%), and coesite (up to 5.5%). Hematite (up to 5.8%) is present only in the upper part of the profile.

4. Discussion

Past pollution and environmental changes have been investigated using mountain peatlands for example in Etang de la Gruyère (1005 m a.s.l.) in Switzerland (Shotyk, 1996; Shotyk et al., 1998, 2002), Black Forest (1044 m a.s.l.), Harz Mts (820 m a.s.l.) (Kempter and Frenzel, 2000; Aubert et al., 2006; Gałka et al., 2019) in Germany, Kovářská Bog (870 m a.s.l.; Bohdálková et al., 2018), Bozi Dar (993–1035 m a.s.l.; Veron et al., 2014) Ore Mountains in the Czech Republic, Motianling (about 1700 m a.s.l.) in China (Bao et al., 2010, 2012, 2015), French Pyrenees (1640 m a.s.l.; Hansson et al., 2017). Some of these peatlands are located on medium altitudes and have received dust mostly from vertical transport because of orographic
barrier of higher summits, which block more lateral transport of particles. The obtained data reveal that Śnieżka shows a more lateral pattern of dust distribution due to the fact of being the highest summit in this part of Europe (see Fig. 1). The detected signals are defined to be more regional due to the fact of being recognized in other countries (comp. discussion below).

### 4.1. Steel factories signal

An increase in Cr and Ni accumulation was described in peatlands from the United States (Cole et al., 1990), Poland (De Vleeschouwer et al., 2009), Belgium (Allan et al., 2013), the Czech Republic (Böhdařová et al., 2018) and Germany (Galka et al., 2019). A distinguishable signal around 1904 ± 15 in Sn1 and from

### Table 3
Pb isotopic signature of peat samples, Karkonosze rocks, aircraft fuel, lignite from mine Turów and fly ash from power plant Turów.

<table>
<thead>
<tr>
<th>Sample</th>
<th>$^{206}\text{Pb}/^{204}\text{Pb}$</th>
<th>$\alpha$</th>
<th>$^{208}\text{Pb}/^{204}\text{Pb}$</th>
<th>$\alpha$</th>
<th>$^{206}\text{Pb}/^{204}\text{Pb}$</th>
<th>$\alpha$</th>
<th>$^{208}\text{Pb}/^{207}\text{Pb}$</th>
<th>Age (cal AD)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sn1 10</td>
<td>38.03005</td>
<td>0.0054</td>
<td>15.62032</td>
<td>0.0019</td>
<td>18.10171</td>
<td>0.002</td>
<td>2.102277</td>
<td>2008</td>
</tr>
<tr>
<td>Sn1 18</td>
<td>38.02355</td>
<td>0.0069</td>
<td>15.62062</td>
<td>0.0018</td>
<td>18.09471</td>
<td>0.0029</td>
<td>2.102677</td>
<td>1997</td>
</tr>
<tr>
<td>Sn1 23</td>
<td>38.18731</td>
<td>0.0037</td>
<td>15.68999</td>
<td>0.0016</td>
<td>18.11321</td>
<td>0.0016</td>
<td>2.10821</td>
<td>1989</td>
</tr>
<tr>
<td>Sn1 25</td>
<td>38.03518</td>
<td>0.008</td>
<td>15.62656</td>
<td>0.0062</td>
<td>18.04409</td>
<td>0.0076</td>
<td>2.107892</td>
<td>1985</td>
</tr>
<tr>
<td>Sn1 28</td>
<td>38.18539</td>
<td>0.011</td>
<td>15.603</td>
<td>0.0042</td>
<td>18.19608</td>
<td>0.0056</td>
<td>2.098303</td>
<td>1978</td>
</tr>
<tr>
<td>Sn1 29</td>
<td>38.0967</td>
<td>0.0054</td>
<td>15.60748</td>
<td>0.0047</td>
<td>18.12604</td>
<td>0.0056</td>
<td>2.101796</td>
<td>1976</td>
</tr>
<tr>
<td>Sn1 30</td>
<td>38.13942</td>
<td>0.005</td>
<td>15.64146</td>
<td>0.0002</td>
<td>18.15352</td>
<td>0.0023</td>
<td>2.104438</td>
<td>1973</td>
</tr>
<tr>
<td>Sn1 33</td>
<td>38.08321</td>
<td>0.0062</td>
<td>15.5768</td>
<td>0.0024</td>
<td>18.16475</td>
<td>0.0029</td>
<td>2.096501</td>
<td>1962</td>
</tr>
<tr>
<td>Sn1 37</td>
<td>38.44224</td>
<td>0.0047</td>
<td>15.6313</td>
<td>0.0019</td>
<td>18.34241</td>
<td>0.0023</td>
<td>2.095801</td>
<td>1946</td>
</tr>
<tr>
<td>Sn1 45</td>
<td>38.33044</td>
<td>0.0034</td>
<td>15.67904</td>
<td>0.0001</td>
<td>18.23704</td>
<td>0.0009</td>
<td>2.107197</td>
<td>1911</td>
</tr>
<tr>
<td>Sn1 49</td>
<td>38.34988</td>
<td>0.0034</td>
<td>15.6884</td>
<td>0.0013</td>
<td>18.18967</td>
<td>0.0014</td>
<td>2.108312</td>
<td>1882</td>
</tr>
<tr>
<td>Sn1 53</td>
<td>38.5393</td>
<td>0.0022</td>
<td>15.69733</td>
<td>0.00073</td>
<td>18.38912</td>
<td>0.00069</td>
<td>2.095686</td>
<td>1829</td>
</tr>
<tr>
<td>Sn1 56</td>
<td>38.52527</td>
<td>0.0056</td>
<td>15.65633</td>
<td>0.0064</td>
<td>18.39197</td>
<td>0.007</td>
<td>2.0947</td>
<td>1797</td>
</tr>
<tr>
<td>Sn1 62</td>
<td>38.55978</td>
<td>0.0073</td>
<td>15.70587</td>
<td>0.0026</td>
<td>18.3789</td>
<td>0.0028</td>
<td>2.097991</td>
<td>1739</td>
</tr>
<tr>
<td>Sn1 72</td>
<td>38.05277</td>
<td>0.0021</td>
<td>15.83808</td>
<td>0.00083</td>
<td>18.19791</td>
<td>0.00083</td>
<td>2.099089</td>
<td>2003.2</td>
</tr>
<tr>
<td>Sn1 8</td>
<td>38.22255</td>
<td>0.0024</td>
<td>15.6253</td>
<td>0.0008</td>
<td>18.21462</td>
<td>0.0008</td>
<td>2.098391</td>
<td>2002.1</td>
</tr>
<tr>
<td>Sn1 9</td>
<td>37.91564</td>
<td>0.0033</td>
<td>15.58858</td>
<td>0.0012</td>
<td>18.11649</td>
<td>0.0014</td>
<td>2.098289</td>
<td>1998.9</td>
</tr>
<tr>
<td>Sn1 11</td>
<td>37.96922</td>
<td>0.0031</td>
<td>15.58799</td>
<td>0.0011</td>
<td>18.08258</td>
<td>0.0001</td>
<td>2.099691</td>
<td>1989.3</td>
</tr>
<tr>
<td>Sn1 13</td>
<td>38.24189</td>
<td>0.0035</td>
<td>15.66424</td>
<td>0.0015</td>
<td>18.1456</td>
<td>0.0017</td>
<td>2.10751</td>
<td>1984.8</td>
</tr>
<tr>
<td>Sn1 15</td>
<td>38.05584</td>
<td>0.0024</td>
<td>15.59224</td>
<td>0.0009</td>
<td>18.12474</td>
<td>0.0011</td>
<td>2.099552</td>
<td>1973.3</td>
</tr>
<tr>
<td>Sn1 17</td>
<td>38.38368</td>
<td>0.0032</td>
<td>15.68434</td>
<td>0.0012</td>
<td>18.2203</td>
<td>0.0014</td>
<td>2.10661</td>
<td>1967.7</td>
</tr>
<tr>
<td>Sn1 21</td>
<td>38.21799</td>
<td>0.0031</td>
<td>15.62475</td>
<td>0.0001</td>
<td>18.2114</td>
<td>0.0001</td>
<td>2.09851</td>
<td>1966.6</td>
</tr>
<tr>
<td>Sn1 27</td>
<td>38.29899</td>
<td>0.0039</td>
<td>15.63554</td>
<td>0.0016</td>
<td>18.2448</td>
<td>0.0018</td>
<td>2.09911</td>
<td>1940</td>
</tr>
<tr>
<td>Sn1 32</td>
<td>38.37868</td>
<td>0.0035</td>
<td>15.62355</td>
<td>0.0013</td>
<td>18.33689</td>
<td>0.0015</td>
<td>2.09291</td>
<td>1925.7</td>
</tr>
<tr>
<td>Sn1 43</td>
<td>38.44178</td>
<td>0.0037</td>
<td>15.63664</td>
<td>0.0014</td>
<td>18.35909</td>
<td>0.0015</td>
<td>2.09391</td>
<td>1894</td>
</tr>
<tr>
<td>Sn1 49</td>
<td>38.3288</td>
<td>0.0019</td>
<td>15.60277</td>
<td>0.00062</td>
<td>18.32207</td>
<td>0.00059</td>
<td>2.097041</td>
<td>1835</td>
</tr>
<tr>
<td>Sn1 53</td>
<td>38.38283</td>
<td>0.0024</td>
<td>15.62275</td>
<td>0.00086</td>
<td>18.35595</td>
<td>0.0011</td>
<td>2.091011</td>
<td>1027</td>
</tr>
</tbody>
</table>

Karkonosze granite

Karkonosze granite

Hornfels (Śnieżka)

Aircraft fuel

Lignite – Turów mine

Lignite – Turów mine

Lignite – Turów mine

Fly ash Turów power plant

Fig. 7. Concentrations and accumulation rates of uranium in Sn1 and Sn2 as well as activity of $^{234}\text{U}$, $^{238}\text{U}$ and $^{234}\text{U}/^{238}\text{U}$ ratio in Sn1.
1922 ± 7–1928 (±6) in Sn2 in the Cr–Ni accumulation profiles is also observed in Śnieżka, with only slight fluctuations in trace and REE accumulation rates (Fig. 5). Ni and Cr are suggested to be immobile in peat bogs (Krachler et al., 2003; Allan et al., 2013), but research indicating mobility also occurred (Nieminen et al., 2002; Ukonnaanaho et al., 2004). Despite the existence of these contradicting conclusions, the repeatable signal in both profiles (Sn1, Sn2) gives us confidence in the possible immobility of those elements. A sharp increase in Cr and Ni in Cowles bog (USA, Northern Indiana), observed in 1928—1942, was explained as booming stainless-steel industry (Cole et al., 1990). The invention of stainless steel in 1912 was a milestone and the boosting of industrial development in many countries (Corker, 2016). In Europe, the leading producer of Cr–Ni steel was Krupp company, the owner of many steel factories, located in Germany (Ruhr area, Kiel) and Poland (presently), for example in Szkłarz (Ger. Glasendorf) (see Fig. 1). The Szkłarz Ni-Steel factory was located about 120 km east of Snieżka, and operated from 1901 to 1920 on both local and imported Ni ores (Furmankiewicz and Krzyżanowski, 2008). In Belgium, the Cr–Ni increase was attributed to industrial activities from the Ruhr region, also belonging to the Krupp company, 300 km from the investigated peatland (Allan et al., 2013). The Ruhr stainless steel industries and their use of Ni–Cr steel alloys promoted the emission of these two metals (Corker, 2016). In the Stowinskie Biota bog (Northern Poland), 7 mg kg⁻¹ was detected in peat layer accumulated in AD 1920, similarly to the NRS bog, but the accumulation rate in Stowinskie Biota was much lower (De Vleeschouwer et al., 2009). The Stowinskie Biota bog is located about 400 km to Kiel and could also be affected by nearby Krupp steel factories. While the exact number of steel factories operating during that time is unknown, the use of primitive technology without particulate-emission control systems has certainly contributed to the elevated Ni and Cr accumulations found in the NRS peat profiles. This is also supported by slag particles found in the Sn1 peat profile at the depth of the highest Cr–Ni concentration and accumulation (Fig. 5). Such particles were not found in the topmost layers. The elevated peak in Cr and Ni in the 1970s and 1980s, together with other investigated elements, is attributed to extensive brown coal combustion (comp. Figs. 4 and 5).

4.2. Uranium processing signal

The Sudetes lie in the northern part of the Bohemian Massif, the most important uranium ore district in Europe with many deposits of various sizes, both in the Czech Republic and bordering eastern Germany. The total historical uranium production is estimated at 350 Mt for the region (OECD-IAEA, 2003). The total production of uranium ores in Germany (Wismut company) from 1946 to 2012 was c. 220 Mt. In Czechoslovakia (presently as the Czech Republic and Slovakia), the total uranium production from 1945 to 2017 was c. 112 Mt (Diehl, 2011). Before becoming a strategic material, uranium was also mined as a by-product during the exploitation of copper, iron, or arsenic ores and used for colouration in glass and porcelain industries (Prochážka et al., 2009). Natural uranium consists of three alpha radioactive isotopes: 99.2745% of 238U, 0.7200% of 235U, and 0.0054% of 234U (Boryto and Skwarzec, 2014). The specific activity of 238U is much lower (1.24 × 104 Bq g⁻¹) compared to 234U (2.30 × 108 Bq g⁻¹) (Brown and Firestone, 1986). A234U/238U ratio close to 1 is typically found in natural samples including rock samples and sediments (Boryto, 2013). The moss Pleurozium schreberi from northern Poland revealed the isotopic ratio also around 1 (Boryto et al., 2017).

A recognizable regional signature is seen in the U record of Śnieżka peat profiles (Fig. 7). In Sn1 and Sn2 profiles, two periods in uranium deposition can be distinguished. The obtained values are two orders of magnitude higher than values reported in the mountainous Etang de la Gruère bog in Switzerland (Krachler and Shotyk, 2004). Moreover, a strong disequilibrium in 234U/238U ratio is observed. From 1872 to 1897, two distinct peaks with value 1.49—1.53 are observed. From 1973 several distinct shifts are observed with value 1.59—1.79, reaching the maximum of 2.3 in 2009. The main natural source of uranium in the air is rock weathering (Boryto and Skwarzec, 2014). The geochemical explanation of radioactivity imbalance is the looser bonds of 234U atoms in mineral structures, making them easier to leach during physicochemical erosion (Flescher and Raabe, 1978). Industrial activities, fossil fuel combustion, phosphate fertilizers in agriculture, and domestic and industrial sewage trigger the increase of uranium concentration and disequilibrium of 234U/238U ratio (Boryto, 2013; Boryto and Skwarzec, 2014).

The elevated values, together with an increased 234U activity coincide with the discovery of uranium and increasing mining and processing of uranium as well as coal combustion. The U peak around 1840—1890 is interpreted here as originating from the glass/porcelain industry. The glassworks used local rock material and charcoal, both enriched in uranium, as well as pure uranium for colouration. This glass industry gained its highest popularity around 1880—1920 (Rene, 2008). Several glassworks existed in the Sudetes and in the South of Bohemia as well as in Germany, Austria, England, and France during that time (Rene, 2008; Prochážka et al., 2009).

The increase in uranium accumulation rate starting from the year 1938 and reaching a maximum in the 1970s—1980s, reflects the pollution from fossil fuels’ burning. In line with other proxies, Uranium was classified as a coalophile element by Ketris and Yudovich (2009). The brown coal from the nearby Turów coal field have twice higher U content (4.4 mg kg⁻¹) than the average for Polish brown (2.2 mg kg⁻¹) and hard (2.0 mg kg⁻¹) coals (Bojakowska et al., 2008). The uranium concentration of Czech brown coal varies from 0.60 to 4.03 mg kg⁻¹ (Bouska and Pesek, 1999). The mining of uranium ores constitutes an additional source of uranium supply to the atmosphere. Taking into account the fact that industrial activities promote the release of 234U to the environment, we propose here that ombrotrophic peatlands can record the influence of humans on the biogeochemical cycle of uranium.

4.3. Chernobyl signal

The Chernobyl accident caused a dramatic release of many radionuclides, which were dispersed throughout Europe. The highest content of radionuclides reached the Belarus boundary, then split up towards Scandinavia and the southwestern part of Europe. It appeared in Poland two days after the accident. The most polluted area was the Opolskie (Southern Poland) and the part of Lower Silesian voidoweship, including Karpacz town, located at the feet of Śnieżka (Łukaszek—Chmielewska et al., 2018). The heterogeneous pattern of radionuclide fallout was caused by diverse weather conditions.

There is a strong increase of 137Cs in the upper 4 cm of Sn1, probably due to plant retention (See Fig. 2 and Table 1). Similarly to K, 137Cs can be uptaken by roots (Gerdol et al., 1994) and accumulated in living plants. This phenomenon was also described elsewhere in peatlands (Mróz et al., 2017; Flakiewicz-Koziet et al., 2014; Rosen et al., 2009). We suspect that the 137Cs from Chernobyl may be present in the top peat layers. The total inventory of 137Cs in Sn1 is 8970 ± 1300 Bq m⁻² (Table 1). The estimated weapons Cs deposition for Poland is 982 Bq m⁻², but the mean total
(including Chernobyl) value for $^{137}$Cs inventory for Poland is 3770 Bq m$^{-2}$ (Stach, 1996). In the Upper Odra valley (SW Poland), the $^{137}$Cs inventory equaled to 5230 Bq m$^{-2}$ (Porzęba and Bluszcz, 2007). The mean inventory value for the other regions of southern Poland (e.g. Opole and Katowice provinces) equals to 11,240 Bq m$^{-2}$ and 6800 Bq m$^{-2}$ respectively (Stach, 1996), but the contribution of $^{137}$Cs of Chernobyl fallout in this region is about 80% of the total $^{137}$Cs fallout. The mean value for $^{137}$Cs inventory for the Tatra National Park (Mietelski et al., 2008) in southern Poland is 7800 Bq m$^{-2}$, and this value is comparable with Sn1. In the Czech Republic, the area from Śnieżka toward Prague was characterised by 3000–10,000 Bq m$^{-2}$, but on NW from Śnieżka only 1000–3000 Bq m$^{-2}$ was detected (Hanslík et al., 2018). The results are in good agreement with previous studies, conducted in higher elevation mountains such as the Tatra Mts.

4.4. Nuclear weapon test signal

There are two main sources of plutonium isotopes: global radioactive atmospheric fallout due to nuclear weapon tests and the Chernobyl accident. Plutonium from the global fallout was spread worldwide, but Pu from the Chernobyl accident was only dispersed in the form of fuel particles and was deposited mostly in the north-eastern and eastern parts of Poland (Mietelski et al., 2016). Non-volatile elements like Pu can be transported on larger aerosol “hot particles” (Cuddihy et al., 1989; Devell et al., 1986). The average deposition of global fallout of $^{239}+^{240}$Pu for Poland was 58 Bq m$^{-2}$ (UNSCEAR, 1982), and about 4% of the activity of $^{239}+^{240}$Pu is due to the average $^{238}$Pu deposition from weapon tests and fallout after the SNAP-9A satellite accidental burn up over the Madagascar. Therefore, we observed lower values for $^{238}$Pu than for $^{239}+^{240}$Pu.

The observed maximum of $^{239}+^{240}$Pu activity concentrations characterize the global fallout peak of 1963 and is commonly used to identify specific horizons in lake sediments, peat, and other deposits, as well as to estimate their accretion rates. The total inventory of $^{239}+^{240}$Pu and $^{238}$Pu is 76.6 ± 7.8 Bq m$^{-2}$ and 2.20 ± 0.71 Bq m$^{-2}$, respectively. The value of $^{239}+^{240}$Pu deposition is slightly higher to the Polish average of 58 Bq m$^{-2}$ predicted by the UNSCEAR (1982). It is, however, in line with values found by Mietelski et al. (2008) in the Kościelska Valley area, Tatra Mts. The $^{238}$Pu/$^{239}+^{240}$Pu activity ratios were calculated for layers with the maximum activity of these isotopes. In the Sn1, the values: 0.018 ± 0.005 to 0.051 ± 0.019 with a mean value of 0.029 ± 0.010 correspond to global fallout, including the SNAP-9A satellite crash (0.03–0.05) and there is no presence of Chernobyl-origin plutonium. The $^{238}$Pu/$^{239}+^{240}$Pu activity ratio for Chernobyl fallout is ~0.50 (Mietelski and Was, 1995) detected only in northern Poland (Mietelski et al., 2016). In the Czech Republic the deposition of plutonium isotopes over former Czechoslovakia (Hölgye and Filipas, 1995) varies from 10.2 to 108.8 Bq m$^{-2}$ for $^{239}+^{240}$Pu and from <0.5 to 6.2 Bq m$^{-2}$ for cumulative $^{238}$Pu (Hölgye and Filipas, 1995; Hölgye and Malý, 2000).

4.5. Coal combustion central Europe signal

In Europe, lead in the atmosphere originated from ore processing, coal combustion, and gasoline usage (Pacyna et al., 2007). The isotopic signature of galena and coal in Central European records are very similar in areas where coal is the main source of lead (Vile et al., 2000; Novak et al., 2003; Cimova et al., 2016). Leaded gasoline, which was an important source of lead in western European countries, is hidden here by coal signatures (Fig. 6). The $^{206}$Pb/$^{207}$Pb signature of aircraft fuel (1.172) taken presently from the main producer of fuel (Laboratory Warter fuels, Płock, Poland) (Table 3), similar to Polish leaded gasoline used in the 1980s (1.174, Yao et al., 2015), is placed within the range of coal signature.

SAP and mullite which form during industrial coal combustion have been deposited on the Śnieżka plateau since 1938 (Figs. 3 and 4), which is later than the electrification of Black Triangle (Krajniak, 2017) and later than the occurrence of fly ash particles in peatlands located close/within coal-based regions (Yang et al., 2001; Śmieja-Król et al., 2019). Lower abundance and technological restrictions (e.g., lower chimneyes resulting in proximal dissemination only) of the earlier times of industrialisation (before ~1950) often give a local/regional signal (Waters et al., 2018) registered in the Śnieżka Mountain only by the occurrence of larger (up to 50 μm) porous aluminosilicates at the depth of 25–35 cm (1912–1942). Such particles were also found in Puścizna Mała (Flakiewicz-Koziet et al., 2011; Śmieja-Król and Flakiewicz-Koziet, 2014) as well as in Bagno Mikołeska in the Silesian Upland, southern Poland (Śmieja-Król and Flakiewicz-Koziet, 2014) and unfound in Mukhrino peatland located in western Siberia (Flakiewicz-Koziet et al., 2016), far from human activity. We hypothesise that they came from closely located emitters via local transport, and because they occurred only before 1950, the main cause was the primitive technology of heating, e.g. from tourist infrastructure at Śnieżka (comp. Kicińska, 2019). However, further studies should be undertaken to support this hypothesis.

The acceleration of coal usage is seen in all proxies shortly after 1970 when the total production of lignite was 260 Gt in Germany (Öko-Institut, 2017), 80 Gt in Czech Republic (Vrablik et al., 2017) and 39 Gt in Poland (Kasztelewicz, 2018). The 1970s are characterised by the highest Pb accumulation rate in both profiles: Sn1 (77 mg m$^{-2}$ yr$^{-1}$) and Sn2 (78 mg m$^{-2}$ yr$^{-1}$) (Fig. 4) comparable to other peatlands from the Black Triangle region (60–65 mg m$^{-2}$ yr$^{-1}$; Novak et al., 2008) or other heavy-industry impacted regions (100 mg m$^{-2}$ yr$^{-1}$; Allan et al., 2013). The highest sum of REE and other elements regarded as lithogenic was also noted for that time and attributed to lignite combustion (comp. with dust flux in Fig. 4., which is calculated based on $^\gamma$REE and La AR as REE representative). It is known that coals contain REE, Th, U, Ti, Zr, and Sc (Bouska and Pesek, 1999; Ketris and Yudovich, 2009; Dolnickova et al., 2012), and that these elements are further concentrated in fly ash (e.g., Vassilev et al., 2001; Jones et al., 2012) emitted in significant amounts during coal combustion. The peak of the booming industry in the 1970s is also recorded by less radiogenic $^{206}$Pb/$^{207}$Pb isotopic ratios (1.161–1.162), distinguished in both profiles (Figs. 4 and 6, Table 3). This is comparable to isotopic signature of the Czech industry (Novak et al., 2003, 2008; Zuna et al., 2011; Bohdílková et al., 2014) and soils in Colditz (eastern Germany within the Black Triangle) (1.161–1.167) (Haack et al., 2003). The isotopic signature of Colditz soil (see Fig. 1) was significantly distinct from the Pb isotopic signature of other locations in western or southern Germany.

The Pb signature is also distinct from the rest of the Polish industry (Cu exploitation, Turów power plant activity) located in the Lower Silesian area (comp. Fig. 6, Table 3, Tyszka et al., 2012). Despite its location in the Black Triangle, the Śnieżka peatland reveals similarities in its Pb isotopic composition with north Poland (De Vleeschouwer et al., 2009), the Czech Republic (Novak et al., 2003), Switzerland (Weiss et al., 1999) and Belgium (De Vleeschouwer et al., 2012). It, however, displays discrepancies with Puścizna Mała - SE Poland (Flakiewicz-Koziet et al., 2018), where distribution patterns are more local due to their foothill character (comp. Fig. 1). The Tatra mountains act here as an effective barrier for long-range transported pollutants. The closeness of the eastern border with various precipitation regimes and wind
directions, as well the burning of peat and coal influenced significantly the isotope ratio of Pb in peatlands from SE Poland, while the Śnieżka peatland displays less radiogenic profiles, suggesting that they receive more long-range transported pollutants, mainly from the west because of the dominant westerlies (comp. Figs. 1 and 6).

A second peak is visible around 1982–1986 and can also be attributed to lignite burning. This period is characterised by the highest production of coal in Poland but also by the highest dust emissions in Saxony (Zimmermann and Bothmer, 2000).

From 1987 onwards, a substantial decrease in all element concentration is observed as a result of reinforced pollution management in Europe. The least radiogenic 206Pb/207Pb isotopic ratio 1.150 (Figs. 4 and 6, Table 3) is comparable to the values of modern aerosols (Bollhöfer et al., 2002) and of unlead gasoline (Bollhöfer and Rosman, 2001), as observed elsewhere in Polish peatlands (De Vleeschouwer et al., 2009).

5. Conclusions

Both peat profiles, sampled in the ombrotrophic mountain peatland, record relatively similar geochemical changes linked to historical industrial activities, despite their different peat accumulation rates. The use of a wide spectrum of geochemical proxies allows distinguishing several broadly recognized sources of pollutants.

1. The Black Triangle industrial activities (the Czech Republic, Germany and Poland) as well as the broad Central Europe coal use caused the most significant changes in the geochemistry of Śnieżka peatland and were the dominant source of pollution and atmospheric emission of most elements such as PHTE (Pb, Zn, Cu, Ni, Cr). All the elements classified as “lithogenic” on a geological viewpoint (Ti, Al, Sc, REE), also increased during 1970’s as a result of the intensive use of coal, pointing out that one should be cautious when using them as reference elements to calculate enrichment factors, as part of these lithogenic elements though are issued from anthropogenic activities.

2. The leaded gasoline signal is hidden by the coal isotopic ratio, which has been abundantly used in central Europe. While the local signal linked to Polish industry is weakened by the elevated altitude, long-range transport is distinguishable, despite the severe pollution of the region. It is confirmed by the size of SAP, which varies from <1–9.5 μm and indicates distal particles.

3. Lead isotopes have indicated a significant contribution of the neighbouring industry to the contamination of Śnieżka.

4. Cr and Ni depositions are strongly influenced by Ni ore smelting and production of stainless steel, also attesting of long-range transport, probably from Germany.

5. The 239+240Pu activity concentrations fingerprint the global fallout peak of nuclear weapon tests, while Chernobyl signal is confirmed only by 137Cs activity.

6. The uranium activity concentration profile displays a complex pattern, and the section corresponding to AD 1828–1938 is clearly disassociated from the topmost part of the profile. The observed disequilibrium in the 234U/238U ratio, exceeding 1.0, indicates an anthropogenic influence on the release of more mobile 234U.

7. Overall, the Sudetes behave like a typical mountain critical zone despite the localisation on lower altitude, and therefore receive more regional, long-range sources of aerosols.

Acknowledgements

This research was funded by the Polish National Centre of Science (NCN) 2011/01/D/ST10/02579 granted to BFK, Jolanta Dopierska (Isotopic Laboratory, UAM) is thanked for Pb isotopes measurements. We would like to thank Danuta Smółka-Danielowska for providing fly ash and Monika Fabińska for lignite samples. Barbara Raniś is thanked for technical help with ash content and bulk density. Marie-Josée Tavella and Aurélie Lanza-nova are thanked for their help in the preparation and measurement of REE at the Service ICMP-MS Observatoire Midi Pyrénées (Toulouse, France). Chuxian Li is thanked for her substantive support.

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.quascirev.2020.106162.

References


Zuna, M., Mihaljević, M., Sebek, O., Ettl, V., Handley, M., Navrátil, T., Goliáš, V., 2011. Recent lead deposition trends in the Czech Republic as recorded by peat bogs and tree rings. Atmos. Environ. 45 (28), 4950–4958.

Łokas, E., Mietelski, J.W., Kleszcz, K., Tomankiewicz, E., 2010. A sequential procedure for determining $^{238}$Pu, $^{239+240}$Pu, $^{241}$Am, $^{90}$Sr, U and Th activities in soils and peats from Spitsbergen. Nukleonika 55, 195–199.
